

# Chemical Performance of PVC

## Abbreviations

S Satisfactory Resistance

L Limited Resistance

U Unsatisfactory Resistance

dil.sol. dilute aqueous solution at a concentration equal to or less than 10%

sol. Aqueous solution at a concentration greater than 10% but not saturated

sat.sol. saturated aqueous solution prepared at 20°C

tg-g technical grade, gas

tg-l technical grade, liquid

tg-s technical grade, solid

work.sol. working solution of the concentration usually used in the industry concerned

susp. Suspension of solid in a saturated solution at 20°C

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Acetaldehyde	CH <sub>3</sub> CHO	20	40	U
		60		U
Acetic acid  -glacial	CH <sub>3</sub> COOH	20	up to 10	S
		60		S
		20	10 to 50	S
		60		L
		20	>96	U
		60		U
Acetic anhydride	(CH <sub>3</sub> CO) <sub>2</sub> O	20	100	U
		60		U
Acetone	CH <sub>3</sub> COCH <sub>3</sub>	20	10	U
		60		U
		20	100	U
		60		U
Acetonitrile		20		U
		60		U
Acetophenone	CH <sub>3</sub> COC <sub>6</sub> H <sub>5</sub>	20	tg-s	U
		60		U
Acetyl nitrile		20		U
		60		U
Acetylene	C <sub>2</sub> H <sub>2</sub>	20	tg-g	S
		60		S
Acrylic acid ethyl ester		20		U
		60		U
Acrylonitrile	CH <sub>2</sub> CHCN	20	technically pure	U
		60		
Adipic acid	(CH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H) <sub>2</sub>	20	sat. sol.	S
		60		L
Air		20	tg-g	S
		60		S
Allyl alcohol	CH <sub>2</sub> CHCH <sub>2</sub> OH	20	tg-l	L
		60		U

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Allyl chloride		20	sat. sol	U
		60		U
Alum (Aluminium potassium sulphate)	$Al_2(SO_4)_3 \cdot K_2SO_4 \cdot nH_2O$	20	sat. sol	S
		60		S
Aluminium -chloride	$AlCl_3$	20	sat. sol.	S
		60		S
-fluoride	$AlF_3$	20	susp.	S
		60		S
-hydroxide	$Al(OH)_3$	20	susp.	S
		60		S
-nitrate	$Al(NO_3)_3$	20	sat. sol.	S
		60		S
-oxychloride		20	susp.	S
		60		S
-sulphate	$Al_2(SO_4)_3$	20	sat. sol.	S
		60		S
Ammonia aqueous dry gas  liquid	$NH_3$	20	sat. sol.	S
		60		S
		20	tg-g	S
		60		S
20	tg-l	L		
60		U		
Ammonium -acetate	$CH_3COONH_4$	20		S
		60		S
-alum		20		S
		60		S
-benzoate		20		S
		60		
-bifluoride		20		S
		60		S
-bisulphate		20		S
		60		S
-carbonate	$(NH_4)_2CO_3$	20	sat. sol.	S
		60		S
-chloride	$NH_4Cl$	20	sat. sol.	S
		60		S
-dichromate		20		S
		60		S
-fluoride	$NH_4F$	20	25	S
		60		L
-hydrogen carbonate	$NH_4HCO_3$	20	sat. sol.	S
		60		S
-hydroxide	$NH_4(OH)$	20	35 m/v sol.	S
		60		S
-nitrate	$NH_4NO_3$	20	sat. sol.	S
		60		S
-persulphate	$(NH_4)_2S_2O_8$	20	sat. sol.	S
		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
<b>Ammonium</b> -phosphate dibasic -phosphate meta -phosphate tri -sulphate -sulphide -thiocyanate -zinc chloride	$\text{NH}_4(\text{HPO}_4)_2$	20		S
		60		S
	$(\text{NH}_4)_4\text{P}_4\text{O}_{12}$	20	sat. sol.	S
		60		S
	$(\text{NH}_4)_2\text{HPO}_4$	20		S
		60		S
	$(\text{NH}_4)_2\text{SO}_4$	20	sat. sol.	S
		60		S
$(\text{NH}_4)_2\text{S}$	20	sat. sol.	S	
	60		S	
-thiocyanate		20	sat. sol.	S
		60		S
-zinc chloride		20		S
		60		S
<b>Amyl acetate</b>	$\text{CH}_3\text{CO}_2\text{CH}_2(\text{CH}_2)_3\text{CH}_3$	20 60	tg-l	U U
<b>Amyl alcohol</b>	$\text{CH}_3(\text{CH}_2)_3\text{CH}_2\text{OH}$	20 60	tg-l	S L
<b>Amyl chloride</b>	$\text{CH}_3(\text{CH}_2)_3\text{CH}_2\text{Cl}$	20 60	tg-l	U U
<b>Aniline</b> -chlorohydrate -hydrochloride -sulphate	$\text{C}_6\text{H}_5\text{NH}_2$	20	sat. sol.	U
		60	or tg-l	U
	$\text{C}_6\text{H}_5\text{NH}_2\text{HCl}$	20		U
		60		U
-hydrochloride		20	sat. sol.	U
		60		U
-sulphate		20		U
		60		U
<b>Anthraquinone</b>		20		S
		60		U
<b>Anthraquinone sulphonic acid</b>		20	susp.	S
		60		S
<b>Antimony chloride</b>	$\text{SbCl}_3$	20	sat. sol.	S
		60		S
<b>Aqua regia</b>	$\text{HCl} + \text{HNO}_3$	20		U
		60		U
<b>Arsenic acid</b>	$\text{H}_3\text{AsO}_4$	20	sat. sol. or	S
		60	weak conc.	L
<b>Aryl sulphonic acids</b>		20		S
		60		U
<b>Barium</b> -bromide -carbonate -chloride -hydroxide -nitrate	$\text{BaBr}_2$	20	sat. sol	S
		60		S
	$\text{BaCO}_3$	20	susp.	S
		60		S
	$\text{BaCl}_2$	20	sat. sol.	S
60			S	
$\text{Ba(OH)}_2$	20	sat. sol.	S	
	60		S	
$\text{Ba(NO}_3)_2$		20		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Barium -sulphate -sulphide	BaSO <sub>4</sub>	20 60	susp.	S S
	BaS	20 60	sat. sol.	S S
Beer		20 60	work sol.	S S
Benzaldehyde	C <sub>6</sub> H <sub>5</sub> CHO	20 60		U U
Benzalkonium chloride		20		S
Benzene	C <sub>6</sub> H <sub>6</sub>	20 60	tg-l	U U
Benzoic acid	C <sub>6</sub> H <sub>5</sub> COOH	20 60	sat. sol.	L U
Benzoyl chloride		20	tg-l	U
Benzyl acetate		20 60		U U
Bismuth carbonate		20 60	sat. sol.	S S
Boric acid	H <sub>3</sub> BO <sub>3</sub>	20 60	sat. sol.	S L
Boron trifluoride	BF <sub>3</sub>	20	sat. sol.	S
Brine		20 60	work sol.	S S
Bromic acid	HBrO <sub>3</sub>	20 60	10	S S
Bromine	Br <sub>2</sub>	20 60	tg-g	U U
		20 60	tg-l	U U
		20 60	trace	L U
Bromobenzene		20 60		U U
Bromoethane		20 60	tg-l	U U
Bromotoluene		20 60		U U
Butadiene	C <sub>4</sub> H <sub>6</sub>	20 60	tg-g	S S
Butane	C <sub>4</sub> H <sub>10</sub>	20 60	tg-g	S S
Butanediols	CH <sub>3</sub> CH <sub>2</sub> CHOHCH <sub>2</sub> OH	20 60	10	S U
		20 60	conc.	L U
Butanols (butyl alcohols)	C <sub>4</sub> H <sub>9</sub> OH	20 60	tg-l	S L
Butyl acetate	CH <sub>3</sub> CO <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	20 60	tg-l	U U
Butylene glycol	C <sub>4</sub> H <sub>6</sub> (OH) <sub>2</sub>	60	100	L

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Butyl mercaptan		20		U
		60		U
Butylphenols	C <sub>4</sub> H <sub>9</sub> C <sub>6</sub> H <sub>4</sub> OH	20	sat. sol.	U
		60		U
Butyl phthalate		20	tg-l	U
		60		U
Butylstearate		20		S
Butynediol		20		S
		60		U
Butyric acid	C <sub>2</sub> H <sub>5</sub> CH <sub>2</sub> COOH	20	20	S
		60		U
		20	tg-l	U
		60		U
Cadmium cyanide		20		S
		60		S
Calcium -carbonate -chlorate  -chloride  -hydrogen sulphide (calcium bisulphide) -hydrogen sulphite (calcium bisulphite) -hydroxide  -hypochlorite  -nitrate  -sulphate  -sulphide	CaCO <sub>3</sub>	20	susp.	S
		60		S
	CaCHCl	20	sat. sol.	S
		60		S
	CaCl <sub>2</sub>	20	sat. sol.	S
		60		S
	Ca(HS) <sub>2</sub>	20	sol.	S
		60		S
	Ca(HSO <sub>3</sub> ) <sub>2</sub>	20		S
		60		S
	Ca(OH) <sub>2</sub>	20	sat. sol.	S
		60		S
	Ca(OCl) <sub>2</sub>	20	sat. sol.	S
		60		S
Ca(NO <sub>3</sub> ) <sub>2</sub>	20	sat. sol.	S	
	60		S	
CaSO <sub>4</sub>	20	susp.	S	
	60		S	
CaS	20	sat. sol.	S	
	60		S	
Carbitol		20		S
Carbon dioxide (gas) (aqueous)	CO <sub>2</sub>	20	tg-g	S
		60		S
		20	sat. sol.	S
		60		S
Carbon disulphide	CS <sub>2</sub>	20	tg-l	U
		60		U
Carbon monoxide	CO	20	tg-g	S
		60		S
Carbon tetrachloride Carbon tetrachloride	CCl <sub>4</sub>	20	tg-l	U
		60		U
Carbonic acid (aqueous) (dry)	H <sub>2</sub> CO <sub>3</sub>	20	sat. sol.	S
		60		S
		20	100	S
		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Carbonic acid (aqueous) (wet)		20		S
		60		L
Castor oil		20		S
		60		S
Caustic potash		20		S
		60		S
Cellosolve (2-ethoxyethanol)		20		S
		60		U
Cellosolve acetate		20		S
Chloral hydrate		20		S
		60		S
Chloramine		20	dil. sol.	S
Chloric acid	HClO <sub>3</sub>	20	20	S
		60		L
Chlorine -dry gas	Cl <sub>2</sub>	20	10	S
		60		L
		20	100	L
		60		U
Chloroacetic acid	ClCH <sub>2</sub> COH	20	sol.	S
		60		L
Chloroacetyl chloride		20		S
Chlorobenzene		20	tg-l	U
		60		U
Chloroform	CHCl <sub>3</sub>	20	tg-l	U
		60		U
Chloropicrin		20		U
Chloropropanes		20	tg-l	U
		60		U
Chlorosulphonic acid	ClHSO <sub>3</sub>	20	tg-s	L
		60		U
Chrome alum	KCr(SO <sub>4</sub> ) <sub>2</sub>	20	sol.	S
		60		S
Chromic acid (plating soln)	CrO <sub>3</sub> + H <sub>2</sub> O	20	10	S
		60		
		20	30	S
		60		
20	50	S		
60		L		
20	sat. sol.	S		
Chromic solution	CrO <sub>3</sub> + H <sub>2</sub> O + H <sub>2</sub> SO <sub>4</sub>	20	50/35/15	S
		60		L
Citric acid	C <sub>3</sub> H <sub>4</sub> (OH)(CO <sub>2</sub> H) <sub>3</sub>	20	sat. sol.	S
		60		S
Copper -carbonate	CuCO <sub>3</sub>	20		S
		60		S
Copper -chloride	CuCl <sub>2</sub>	20	sat. sol.	S
		60		S
Copper -cyanide	CuCN <sub>2</sub>	20	sat. sol.	S
		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Copper -fluoride  -hypochlorite  -nitrate  -sulphate	CuF <sub>2</sub>	20		S
		60		S
	Cu(OCl) <sub>2</sub>	20		S
		60		S
Cu(NO <sub>3</sub> ) <sub>2</sub>	20	sat. sol.	S	
	60		S	
CuSO <sub>4</sub>	20	sat. sol.	S	
	60		S	
Cottonseed oil		20	work	S
		60	sol.	S
Creosote		20		U
		60		U
Cresol	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> OH	20	≤90	L
		60		U
		20	≥90	U
60		U		
Cresylic acid	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> COOH	20	50	L
		60		U
Crotonaldehyde		20	sat. sol. or	U
		60	tg-l	U
Crude oil		20	tg-l	S
		60		S
Cyclanone		20		S
		60		S
Cyclohexane	C <sub>6</sub> H <sub>12</sub>	20		U
		60		U
Cyclohexanol		20	sat. sol. or	U
		60	tg-s	U
Cyclohexanone	C <sub>6</sub> H <sub>10</sub> O	20	tg-l	U
		60		U
Cyclohexyl alcohol		20		U
		60		U
DDT		20		U
		60		U
Detergents (synthetic)		20	dil	S
		60		S
Developers (photographic)		20	work	S
		60	sol.	S
Dextrin	C <sub>6</sub> H <sub>12</sub> OCH <sub>2</sub> O	20	sol.	S
		60		L
Dextrose		20	sol.	S
		60		S
Diacetone alcohol		22		S
Diazo salts		20		S
		60		S
Dibutoxyethyl phthalate		20		U
		60		U
Dibutyl phthalate	C <sub>6</sub> H <sub>4</sub> (CO <sub>2</sub> C <sub>4</sub> H <sub>9</sub> ) <sub>2</sub>	20		U
		60		U
Dibutyl sebacate		20		S
		60		U

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Dichloroacetic acid	Cl <sub>2</sub> CHCOOH	20 60	tg-l	U U
Dichlorobenzene		20 60	tg-l	U U
Dichloroethane (ethylene dichloride)	CH <sub>2</sub> ClCH <sub>2</sub> Cl	20 60	tg-l	U U
Dichloroethylene	ClCH <sub>2</sub> Cl	20 60	tg-l	U U
Diesel fuels		20 60		S S
Diethyl ether	C <sub>2</sub> H <sub>5</sub> OC <sub>2</sub> H <sub>5</sub>	20 60		U U
Diethyl sulphate (ethyl sulphate)	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> SO <sub>4</sub>	20 60		U U
Diglycolic acid	(CH <sub>2</sub> ) <sub>2</sub> O(CO <sub>2</sub> H) <sub>2</sub>	20 60		S L
Dimethylamine	(CH <sub>3</sub> ) <sub>2</sub> NH	20 60	100	L U
Dimethyl formamide		20 60		U U
Dimethylhydrazine		20 60		U U
Dimethyl sulphate (methyl sulphate)	(CH <sub>3</sub> ) <sub>2</sub> SO <sub>4</sub>	20 60		S U
Diocetyl phthalate		20 60	tg-l	U U
Dioxane		20 60	tg-l	U U
Diphenyl ether		20 60		U U
Dodecanoic acid (lauric acid)		20 60		S S
Emulsions (photographic)		20 60	work sol.	S S
Ethanol (ethyl alcohol)	CH <sub>3</sub> CH <sub>2</sub> OH	20 60	tg-l	S L
Ethers		20 60		U U
Ethyl -acetate -acrylate  -chloride  -chloroacetate  -ether	CH <sub>3</sub> CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub>	20 60	tg-l	U U
		20 60	tg-l	U U
	CH <sub>3</sub> CH <sub>2</sub> Cl	20 60	tg-g	U U
		20 60		U U
	CH <sub>3</sub> CH <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	20 60	tg-l	U U
		20 60		U U
Ethylene -chlorohydrin	ClCH <sub>2</sub> CH <sub>2</sub> OH	20 60	tg-l	U U



Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Ethylene -dibromide		20		U
		60		U
Ethylene -glycol (ethanediol)	HOCH <sub>2</sub> CH <sub>2</sub> OH	20	tg-l	S
		60		S
-oxide (oxiran)		20		U
		60		U
Fatty acids		20		S
		60		S
Ferric -acetate	Fe(CH <sub>3</sub> COO) <sub>3</sub>	20		S
		60		U
-chloride	FeCl <sub>3</sub>	20	sat. sol.	S
		60		S
-hydroxide	Fe(OH) <sub>3</sub>	20		S
		60		S
-nitrate	Fe(NO <sub>3</sub> ) <sub>3</sub>	20	sat. sol.	S
		60		S
-sulphate	Fe(SO <sub>4</sub> ) <sub>3</sub>	20	sat. sol.	S
		60		S
Ferrous -chloride	FeCl <sub>2</sub>	20	sat. sol.	S
		60		S
-hydroxide	Fe(OH) <sub>2</sub>	20		S
-nitrate	FeNO <sub>3</sub>	20		S
-sulphate	FeSO <sub>4</sub>	20	sat. sol.	S
		60		S
Fixing soln. (photographic)		20		S
		60		S
Fluoboric acid		20		S
		60		S
Fluorine	F <sub>2</sub>	20	tg-g wet or dry	U
		60		U
Fluosilic acid	HSiF <sub>6</sub>	20	sat. sol.	S
		60		S
Formaldehyde	HCOH	20	30-40%	S
		60		S
Formic acid	HCOOH	20	10	S
		60		S
		20	25	S
		60		L
		20	50	S
		60		L
20	100	S		
60		U		
Fructose		20		S
		60		S
Fuel oil		20		S
		60		S
Furfuraldehyde (furfural)		20		U
		60		U

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Furfuryl alcohol	C <sub>5</sub> H <sub>3</sub> OCH <sub>2</sub> OH	20 60	tg-l	U U
Gas (manufactured)		20 60	tg-g	S L
Gas (natural,wet/dry)		20	tg-g	S
Gasoline (fuel)		20 60	work sol.	S S
Gelatine		20 60	sol.	S S
Glucose	C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	20 60	sol.	S S
Glycerine	HOCH <sub>2</sub> CHOHCH <sub>2</sub> OH	20 60	tg-l	S S
Glycolic acid	HOCH <sub>2</sub> COOH	20 60	30	S S
Heptane	C <sub>7</sub> H <sub>16</sub>	20 60	tg-l	S U
Hexadecanol (cetyl alcohol)		20 60	work sol.	S S
Hexane	C <sub>6</sub> H <sub>14</sub>	20 60		S L
Hexanol (hexyl alcohol)		20 60	tg-l	S S
Hydrazine		20 60	97	U U
Hydrobromic acid	HBr	20 60 20 60	up to 20 50	S L S L
Hydrochloric acid	HCl	20 60 20 60	≤25 ≤37	S L S S
Hydrocyanic acid	HCN	20 60	10	S S
Hydrofluoric acid	HF	20 60 20 60 20 60	up to 10 40 60	S S L U L U
Hydrogen	H <sub>2</sub>	20 60		S S
-peroxide	H <sub>2</sub> O <sub>2</sub>	20 60 20 60 20 60	12 30 50 90	S S S S S S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC	
Hydrogen-sulphide	H <sub>2</sub> S	20	tg-g	S	
		60		S	
Hydroquinone (quinol)		20	sat. sol.	S	
		60		S	
Hydrosulphite		20	≤10	S	
		60		L	
Hydroxylamine sulphate	(H <sub>2</sub> NOH) <sub>2</sub> H <sub>2</sub> SO <sub>4</sub>	20	12	S	
		60		S	
Hydrochlorous acid		20		L	
Hypochlorite		20		S	
Hypochlorous acid		20		S	
		60		S	
Iodine (soln in potassium iodide) (soln in alcohol)	I <sub>2</sub>	20	sat. sol.	U	
		60		U	
		20	tg-l	U	
		60		U	
Isobutyl alcohol		20	tg-l	S	
		60		S	
Iso-octane (2,2,4-trimethylpentane)	C <sub>8</sub> H <sub>18</sub>	20		S	
		60		U	
Isophorone		20		U	
		60		U	
Isopropyl-alcohol-ether	(CH <sub>3</sub> ) <sub>2</sub> CHOH	20	tg-l	S	
		60		S	
	(CH <sub>3</sub> ) <sub>2</sub> CHOCH(CH <sub>3</sub> ) <sub>2</sub>	20		L	
		60		U	
Kerosene		20		S	
		60		S	
Lactic acid	CH <sub>3</sub> CHOHCOOH	20	10	S	
		60		L	
		20	10 to 90	L	
		60		U	
Latex		20		S	
		60		S	
Lauryl chloride		20		S	
		60		U	
Lead-acetate-arsenate-chloride-nitrate-sulphate	Pb(CH <sub>3</sub> COO) <sub>2</sub>	20	dil. or sat. sol.	S	
		60		S	
		20		S	
		60		S	
	PbCl <sub>2</sub>	20		S	
		60		S	
	PbNO <sub>3</sub>	20		S	
		60		S	
	PbSO <sub>4</sub>	20		S	
		60		S	
	Linoleic acid		20		S
			60		S
Linoleic oil		20		S	
		60		S	

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Linseed oil		20	work sol.	S
		60		L
Lithium bromide		20		S
		60		S
Magnesium -carbonate -chloride -citrate -hydroxide -nitrate -sulphate	MgCO <sub>3</sub>	20	susp.	S
		60		S
	MgCl <sub>2</sub>	20	sat. sol.	S
		60		S
		20		S
		60		S
	Mg(OH) <sub>2</sub>	20	sat. sol.	S
60		S		
MgNO <sub>3</sub>	20	sat. sol.	S	
	60		S	
MgSO <sub>4</sub>	20	sat. sol.	S	
	60		S	
Maleic acid	COOHCHCOOH	20	25	S
		60		S
		20	50	S
		60		S
	20	sat. sol.	S	
	60		L	
Malic acid	CH <sub>2</sub> CHOH(COOH) <sub>2</sub>	20	sol. or sat. sol.	S
		60		S
Manganese -chloride -sulphate		20		S
		60		S
		20		10/20 or sat.
60	S			
Mercuric -chloride -cyanide	HgCl <sub>2</sub>	20	sat. sol.	S
		60		S
	HgCN <sub>2</sub>	20	sat. sol.	S
60	S			
Mercurous nitrate	HgNO <sub>3</sub>	20		S
		60		S
Mercury	Hg	20	tg-l	S
		60		S
Mesityl oxide		20		U
		60		U
Methoxyethyl oleate		20		S
		60		S
Methyl -acetate -alcohol (methanol)  -bromide (bromomethane) -cellosolve	CH <sub>3</sub> COOCH <sub>3</sub>	20	tg-l	U
		60		U
	CH <sub>3</sub> OH	20	5	S
		60		S
		20		tg-l
	60	L		
CH <sub>3</sub> Br	20		U	
	60		U	
	20		U	
	60			

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC	
<b>Methyl -chloride (chloromethane)</b>	CH <sub>3</sub> Cl	20		U	
		60		U	
	<b>-ethyl ketone</b>	CH <sub>3</sub> COCH <sub>2</sub> CH <sub>3</sub>	20	tg-l	U
			60		U
	<b>-glycol</b>		20		S
			60		S
	<b>-isobutyl ketone</b>		20	tg-l	U
60				U	
<b>-methacrylate</b>		20	tg-l	U	
		60		U	
<b>-salicylate</b>		20		S	
<b>Methylamine</b>	CH <sub>3</sub> NH <sub>2</sub>	20	32	L	
<b>Methylated spirits</b>		20		S	
		60		L	
<b>Methylcyclohexanone</b>		20		U	
		60		U	
<b>Methylene -bromide</b>	CH <sub>2</sub> Br <sub>2</sub>	20		U	
		60		U	
	<b>-chloride</b>	CH <sub>2</sub> Cl <sub>2</sub>	20	tg-l	U
			60		U
<b>-chlorobromide</b>		20		U	
		60		U	
<b>-iodine</b>		20		U	
		60		U	
<b>Methylsulphoric acid</b>	CH <sub>3</sub> COOSO <sub>4</sub>	20	50/100	S	
		60		L	
<b>Mineral oils</b>		20	work	S	
		60	sol.	S	
<b>Molasses</b>		20	work	S	
		60	sol.	L	
<b>Motor oils</b>		20		S	
		60		S	
<b>Muriatic acid</b>		20		S	
		60		S	
<b>Naphtha</b>		20	work. sol.	U	
		60		U	
<b>Naphthalene</b>	C <sub>10</sub> H <sub>8</sub>	20		U	
		60		U	
<b>Natural gas</b>		20		S	
		60		S	
<b>Nickel -acetate</b>	Ni(CH <sub>3</sub> COO) <sub>2</sub>	20		S	
		60		S	
	<b>-chloride</b>	NiCl <sub>2</sub>	20	sat. sol.	S
			60		S
<b>-nitrate</b>	Ni(NO <sub>3</sub> ) <sub>2</sub>	20	sat. sol.	S	
		60		S	
<b>-sulphate</b>	NiSO <sub>4</sub>	20	sat. sol.	S	
		60		S	
<b>Nicotonic acid</b>		20	susp.	S	
		60		S	

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Nitric acid	HNO <sub>3</sub>	20	up to 45%	S
		60		L
		20	>50%	U
		60		U
Nitrobenzene	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	20	tg-l	U
		60		U
Nitroglycerin		20		U
		60		U
Nitroglycol		20		U
		60		U
Nitromethane		20		L
Nitropropane		20		U
		60		U
Nitrous fumes (moist)		20		L
		60		U
Nitrous oxide	N <sub>2</sub> O	20		S
		60		U
Oils and fats		20	tg-l	S
		60		S
Oleic acid	C <sub>8</sub> H <sub>17</sub> CHCH(CH <sub>2</sub> ) <sub>7</sub> CO <sub>2</sub> H	20	tg-l	S
		60		S
Oleum		20		U
		60		U
Oxalic acid	HO <sub>2</sub> CCO <sub>2</sub> H	20	sat. sol.	S
		60		S
		20	dil. sol.	S
60	L			
Oxygen	O <sub>2</sub>	20	tg-g	S
		60		S
Ozone	O <sub>3</sub>	20	sat. sol.	S
		60		S
Palmitic acid	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>14</sub> COOH	20	10	S
		60		S
		20	70	S
		60		S
Paraffin		20		S
		60		L
-emulsion/oil		20		S
		60		S
Peracetic acid		20		S
		60		U
Perchloric acid	HClO <sub>4</sub>	20	10	S
		60		L
		20	70	L
		60		U
Perphosphate		20		S
Petrol		20		S
		60		U
		25		S
-unrefined		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC	
Petrol/benzene (mixture)		20	80:20	U	
		60		U	
Petroleum spirit (petroleum ether)		20		U	
		60		U	
Petroleum liquifier		22		S	
		60		S	
Petroleum oils		22		S	
		60		U	
Phenol	C <sub>6</sub> H <sub>5</sub> OH	20	1	S	
		20	90	U	
		60		U	
Phenylhydrazine	C <sub>6</sub> H <sub>5</sub> NHNH <sub>2</sub>	20	tg-l	U	
		60		U	
Phenylhydrazine hydrochloride	C <sub>6</sub> H <sub>5</sub> NHNH <sub>3</sub> Cl	20	dil. sol.	U	
		60		U	
Phosgene (gas) (liquid)		20		S	
		60		U	
		20		U	
		60		U	
Phosphine		20	tg-g	S	
		60		S	
Phosphoric -acid	H <sub>3</sub> PO <sub>4</sub>	20	10	S	
		60		S	
		20	25	S	
		60		S	
	-anhydride	P <sub>2</sub> O <sub>5</sub>	20	50	S
			60		S
			20	95	S
			60		S
Phosphorous	P <sub>4</sub>	20		S	
		60		U	
-pentoxide	P <sub>2</sub> O <sub>5</sub>	20		S	
		60		U	
-trichloride	PCl <sub>3</sub>	20	tg-l	U	
		60		U	
Phosphoryl chloride (phosphorus oxychloride)		20	tg-l	U	
		60		U	
Phthalic acid	C <sub>6</sub> H <sub>4</sub> (CO <sub>2</sub> H) <sub>2</sub>	20 60	50	U	
Picric acid:	HO <sub>6</sub> H <sub>2</sub> (NO <sub>2</sub> ) <sub>3</sub>	20	1	S	
		60		S	
		20 60	≥1	U U	
Plating solutions: brass cadmium chromium		20		S	
		60		S	
		20		S	
		60		S	
		20		S	
		60		S	

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
<b>Plating solutions:</b>		20		S
		60		S
copper		20		S
		60		S
gold		20		S
		60		S
indium		20		S
		60		S
lead		20		S
		60		S
nickel		20		S
		60		S
rhodium		20		S
		60		S
silver		20		S
		60		S
tin		20		S
		60		S
zinc		20		S
		60		S
<b>Polyglycol ethers</b>		20		U
		60		U
<b>Potash</b>		20		S
		60		S
<b>Potassium</b>		20		S
		60		S
-alum				
-borate	K <sub>3</sub> BO <sub>3</sub>	20 60	sat. sol.	S S
-bromate	KBrO <sub>3</sub>	20 60	up to 10	S S
-bromide	KBr	20 60	sat. sol.	S S
-carbonate	K <sub>2</sub> CO <sub>3</sub>	20 60	sat. sol.	S S
-chlorate		20 60	sat. sol.	S S
-chloride	KCl	20 60	sat. sol.	S S
-cuprocyanide		20	sat. sol.	S
-chromate	K <sub>2</sub> CrO <sub>4</sub>	20 60	40	S S
-cuprocyanide		20 60	sat. sol.	S
-cyanide	KCN	20 60	sat. sol.	S S
-dichromate (potassium bichromate)	K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	20	40	S
		60		S
		20	sat. sol.	S
-ferricyanide		20	sat. sol.	S
		60		S
-ferrocyanide (potassium hexacyanoferrate (II))	K <sub>4</sub> Fe(CN) <sub>6</sub> ·3H <sub>2</sub> O	20	sat. sol.	S
		60		S



Chemical	Formula	Temp. (°C)	Conc. (%)	PVC	
<b>Potassium</b> -fluoride -hydrogen carbonate (potassium bicarbonate) -hydrogen sulphate (potassium bisulphate) -hydrogen sulphite (potassium bisulphite) -hydroxide  -nitrate  -perborate  -perchlorate  -permanganate  -persulphate  -sulphate  -sulphide  -sulphite  -thiosulphate	KF	20	sat. sol.	S	
		60		S	
			20	sat. sol.	S
			60		S
			20	sat. sol.	S
			60		S
			20	sol.	S
			60		S
	KOH		20	10	S
			60		S
			20	50	S
			60		S
			20	conc.	S
			60		S
	KNO <sub>3</sub>		20	sat. sol.	S
			60		S
	KBO <sub>3</sub>		20		S
			60		S
			20	10	S
			60		S
KMnO <sub>4</sub>		20	10	S	
		60		S	
		20	20	S	
		60		S	
		20	30	S	
		60		S	
K <sub>2</sub> S <sub>2</sub> O <sub>8</sub>		20	sat. sol.	S	
		60		L	
K <sub>2</sub> SO <sub>4</sub>		20	sat. sol.	S	
		60		S	
		20	sat. sol.	S	
		60		S	
		20	sat. sol.	S	
		60		S	
		20	sat. sol.	S	
		60		S	
<b>Propane</b>	C <sub>3</sub> H <sub>8</sub>	20		S	
		60		S	
<b>Propylene</b> -dichloride -oxide		20		U	
		60		U	
		20		U	
		60		U	
<b>Pyridine</b>	CH(CHCH) <sub>2</sub> N	20		U	
		60		U	
<b>Salicylic acid</b>		20	sat. sol.	S	
		60		S	
<b>Sea Water</b>		20		S	
		60		S	
<b>Sewage</b>		20		S	
		60		S	

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
<b>Silicic acid</b>	$H_2SiO_3$	20		S
		60		S
<b>Silver</b> -acetate -cyanide  -nitrate	$AgCH_3COO$	20	sat. sol.	S
		60		S
	$AgCN$	20	sat. sol.	S
		60		S
$AgNO_3$	20	sat. sol.	S	
	60		L	
<b>Soap solutions</b> (aqueous soln.)		20	sol.	S
		60		L
<b>Sodium</b> -acetate -alum	$CH_3COONa$	20		S
		60		S
-antimonate		20	sat. sol.	S
		60		S
-arsenite		20	sat. sol.	S
		60		S
-benzoate		20		S
		60		L
-bicarbonate (hydrogen carbonate)	$NaHCO_3$	20	sat. sol.	S
		60		S
-bichromate (hydrogen chromate)		20		S
		60		S
-bisulphate (hydrogen sulphate)	$NaHSO_4$	20	sat. sol.	S
		60		S
-bisulphite (hydrogen sulphite)	$NaHSO_3$	20	sat. sol.	S
		60		S
-bromide	$NaBr$	20	sat. sol.	S
		60		S
-carbonate	$Na_2CO_3$	20	sat. sol.	S
		60		S
-chlorate	$NaClO_3$	20	sat. sol.	S
		60		S
-chloride	$NaCl$	20	sat. sol.	S
		60		S
-cyanide	$NaCN$	20	sat. sol.	S
		60		S
-dichromate		20		S
		60		S
-ferricyanide		20	sat. sol.	S
		60		S
-ferrocyanide	$Na_4Fe(CN)_6$	20	sat. sol.	S
		60		S
-fluoride	$NaF$	20	sat. sol.	S
		60		S
-hydrogen orthophosphate (di Sodium -)		20		S
		60		S
-hydroxide	$NaOH$	20	1 w/v	S
		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
<b>Sodium -hydroxide</b>		20	10 w/v	S
		60		S
<b>-hypochlorite</b>	NaOCl	20	40 w/v	S
		60		S
<b>-metaphosphate</b>		20	conc.	S
		60		S
<b>-nitrate</b>	NaNO <sub>3</sub>	20	13% Cl	S
		60		L
<b>-nitrite</b>	NaNO <sub>2</sub>	20	sat. sol.	S
		60		S
<b>-perborate</b>	NaBO <sub>3</sub> .H <sub>2</sub> O	20		S
		60		S
<b>-perchlorate</b>		20		S
		60		S
<b>-peroxide</b>		20		S
		60		S
<b>-phosphate di</b>	NaHPO <sub>4</sub>	20		S
		60		S
<b>-phosphate tri</b>	Na <sub>3</sub> PO <sub>4</sub>	20		S
		60		S
<b>-silicate</b>		20	sol.	S
		60		S
<b>-sulphate</b>	Na <sub>2</sub> SO <sub>4</sub>	20	dil. or sat. sol.	S
		60		S
<b>-sulphide</b>	Na <sub>2</sub> S	20	dil.	S
		60		L
<b>-sulphite</b>	NaSO <sub>3</sub>	20	sat. sol.	S
		60		L
<b>-tetraborate (di Sodium-), 'Borax'</b>		20		S
		60		S
<b>-thiosulphate (sodium hyposulphite)</b>	Na <sub>2</sub> S <sub>3</sub> O <sub>3</sub>	20		S
		60		S
<b>Stannic chloride (Tin (IV) chloride)</b>	SnCl <sub>4</sub>	20	sol.	S
		60		S
<b>Stannous chloride (Tin (II) chloride)</b>	SnCl <sub>2</sub>	20	sat. sol.	S
		60		S
<b>Starch</b>		20		S
		60		S
<b>Stearic acid</b>	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>16</sub> CO <sub>2</sub> H	20		S
		60		S
<b>Stoddard solvents</b>		20		U
		60		U
<b>Succinic acid</b>		20		S
		60		S
<b>Sucrose (sugar)</b>		20	aq. sol.	S
		60		S

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Sulphamic acid		20	sol.	S
Sulphite liquors		20 60		S S
Sulphur	S	20 60		S S
Sulphur dioxide (dry)	SO <sub>2</sub>	20 60		S S
(moist)		20 60		S U
(liquid)		20 60		L U
Sulphur trioxide	SO <sub>3</sub>	20 60		S S
Sulphuric acid	H <sub>2</sub> SO <sub>4</sub>	20 60	up to 10	S S
		20 60	15	S S
		20 60	10 to 50	S S
		20 60	50 to 90	S L
		20 60	95	L U
		20 60	98	U U
		20 60	fuming	U U
-nitric aqueous soln.	H <sub>2</sub> SO <sub>4</sub> + HNO <sub>3</sub> + H <sub>2</sub> O	20 60	48/49/3	S L
		20 60	50/50/0	L U
		20 60	#####	S S
Sulphurous acid		20 60	10	S S
		20 60	30	S S
Tallow		20 60		S S
Tannic acid	C <sub>14</sub> H <sub>10</sub> O <sub>9</sub>	20 60	sol.	S S
Tanning extracts		20 60		S S
Tartaric acid	HOOC(CHOH) <sub>2</sub> COOH	20 60	sol. or sat. sol.	S S
Tetrachloroethane	CHCl <sub>2</sub> CHCl <sub>2</sub>	20 60		U U
Tetrachloroethylene (Perchloroethylene)	CCl <sub>2</sub> CCl <sub>2</sub>	20 60		U U
Tetraethyl lead (lead tetraethyl)	Pb(C <sub>2</sub> H <sub>5</sub> ) <sub>4</sub>	20 60	100	S L

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Tetrahydrofuran	C <sub>4</sub> H <sub>8</sub> O	20	tg-l	U
		60		U
Tetrahydronaphthalene (tetralin)		20		U
		60		U
Tetrasodium pyrophosphate		20		S
		60		S
Thionyl chloride	SOCl <sub>3</sub>	20	tg-l	U
		60		U
Thiophene	C <sub>4</sub> H <sub>4</sub> S	20		U
		60		U
Tirpineol		20		S
Titanium tetrachloride		20		U
		60		U
Toluene	C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>	20	tg-l	U
		60		U
Tributyl citrate		20		S
Tributyl phosphate		20		U
		60		U
Trichloroacetic acid	CCl <sub>3</sub> COOH	20	≤50	S
		60		U
Trichlorobenzene		20	work. sol.	U
		60		U
Trichloroethylene	Cl <sub>2</sub> CCHCl	20	tg-l	U
		60		U
Triethanolamine	N(CH <sub>2</sub> CH <sub>2</sub> OH) <sub>2</sub>	20	100	L
		60		U
Triethylamine		20		S
		60		L
Trigol (triethylene glycol)		20		S
		60		
3,4,5,-Trihydroxybenzoic acid (gallic acid)		20		S
		60		S
Trilon		20		U
		60		U
Trimethylamine		20		S
		60		U
Trimethylol propane (2-ethyl-2-hydroxymethylpropanediol)		20	up to 10%	S
		60		L
Trimethyl propane		20		S
		60		L
Trisodium phosphate		20		S
		60		S
Turpentine		20		S
		60		L
Urea	CO(NH <sub>2</sub> ) <sub>2</sub>	20	≤10	S
		60		L
		20	33	S
60	L			

Chemical	Formula	Temp. (°C)	Conc. (%)	PVC
Uric acid	C <sub>5</sub> H <sub>4</sub> N <sub>4</sub> O <sub>3</sub>	20 60	10	S L
Urine		20 60		S
Vegetable oils		20 60		S S
Vinegar		20 60		S S
Vinyl acetate	CH <sub>3</sub> CO <sub>2</sub> CHCH <sub>2</sub>	20 60	tg-l	U U
Water	H <sub>2</sub> O	20 60		S S
Whiskey		20 60	work sol.	S S
White liquor		20 60		S S
Wines and spirits		20 60	work sol.	S S
Xylene	C <sub>8</sub> H <sub>10</sub>	20 60	tg-l	U U
Yeast		20 60	susp.	S L
Zinc -carbonate	ZnCO <sub>3</sub>	20 60	susp.	S S
-chloride	ZnCl <sub>2</sub>	20 60	dil. or sat. sol.	S S
-chromate	ZnCrO <sub>4</sub>	20 60		S S
-cyanide	Zn(CN) <sub>2</sub>	20 60		S S
-nitrate	Zn(NO <sub>3</sub> ) <sub>2</sub>	20 60	sat. sol.	S S
-oxide	ZnO	20 60	susp.	S S
-sulphate	ZnSO <sub>4</sub>	20 60	dil. or sat. sol.	S S

#### Sources for Chemical Resistances of PVC

1. The Water Supply Manual for PVC Pipe Systems, First Edition, Vinidex Pty Limited, 1989
2. Chemical Resistance Guide For Thermoplastic Pipe and Fitting Systems, Vinidex Pty Limited
3. ISO/TR 10358 Technical Report: Plastic Pipes and Fittings-Combined Chemical-resistance Classification Table, First Edition, International Organisation for Standardisation, 1993
4. Chemical Resistance, Volume 1- Thermoplastics, Second Edition, Plastics Design Library, 1994
5. Chemical Resistance Data Sheets, Volume 1-Plastics, Rapra Technology Limited, 1993